

Abstract-21**Investigations on CuO: ZnO composites for photocatalytic performance****Akanksha S Chougale^{1,3}, Snehal S Wagh², Harshad D Shelake³, Habib M. Pathan³ and Dnyaneshwar R Shinde^{1*}**¹ P.D.E.A's. Prof. Ramkrishna More ACS College, Akurdi, Pune, India² Dr. Vishwanath Karad MIT World Peace University, Pune, India³ Advanced Physics Laboratory, Department of Physics, Savitribai Phule Pune University, Pune, India

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Abstract: In recent years, semiconductor photocatalysis has been established to be one of the effective strategies for solving global energy and environmental problems by effective utilization of solar energy. Here we have synthesized CuO: ZnO composites using a simple nontoxic chemical route and explored it as an effective photocatalyst for the degradation of rose Bengal (RB) and methylene blue (M.B.) dyes. The prepared CuO: ZnO composites were investigated for their structural, optical, and morphological properties using XRD analysis, UV spectroscopy, EDS, and SEM. The effect of dye concentration, photocatalyst loading dosage, pH, and degradation time under visible light irradiation has been studied. The results demonstrate that CuO: ZnO composite effectively bleached out MB and RB dyes as compared to bare CuO and ZnO.

Keywords: Copper oxide, Zinc oxide, Photocatalysis, Methylene blue, Rose Bengal, Visible light

Abstract-22**Magnetic hyperthermia with Fe₃O₄ nanoparticles****Amol B. Pandhare^{a,b}, Rajendra P. Patil^{b*}, Sagar D. Delekar^{a*}**^a Department of Chemistry, Shivaji University, Kolhapur, Maharashtra 416 004 MS, India.^b Department of Chemistry, M.H. Shinde Mahavidyalaya, Tisangi, Gaganbavda, Kolhapur, Maharashtra 416 206, MS, India.

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Abstract: Biocompatibility restrictions have limited the use of magnetic nanoparticles for magnetic hyperthermia therapy to iron oxides, namely magnetite (Fe₃O₄) and maghemite (g-Fe₂O₃). However, there is yet another magnetic iron oxide phase that has not been considered so far, despite its unique magnetic properties: Fe₃O₄. Indeed, whereas Fe₃O₄ and g-Fe₂O₃ have a relatively low magnetic coercivity, Fe₂O₃ exhibits a giant coercivity. In this report, the heating power of Fe₂O₃ nanoparticles in comparison with g-Fe₂O₃ nanoparticles of similar size (20 nm) was measured in a wide range of field frequencies and amplitudes, in uncoated and polymer-coated samples. It was found that Fe₃O₄ nanoparticles primarily heat in the low-frequency regime (20–100 kHz) in media whose viscosity is like that of cell cytoplasm. In contrast, g-Fe₂O₃ nanoparticles heat more effectively in the high frequency range (400–900 kHz). Cell culture experiments exhibited no toxicity in a wide range of nanoparticle concentrations and a high internalization rate. In conclusion, the performance of Fe₃O₄ nanoparticles is slightly inferior to that of g-Fe₂O₃ nanoparticles in human magnetic hyperthermia applications. However, these Fe₃O₄ nanoparticles open the way for switchable magnetic heating owing to their distinct response to frequency.

Keywords: - Fe₂O₃ nanoparticles, Biocompatibility, magnetic coercivity

Abstract-23**Studies on Induced Mutation in Barnyard Millet (*Echinochloa esculanta*) L.****Jagtap Bhavana D¹, Danai-Tambhale S. D.²**^{1&2} Department of Botany, Annasaheb Magar Mahavidyalaya, Hadapsar, Pune-411028, Maharashtra

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Abstract: Induced mutation is one of the highly effective and valuable tools in plant breeding techniques. Inducing new genetic recombinant with desired traits could be successful research pure line crop. The ultimate goal of plant breeding in millet is to get maximum grain yield including biomass and the harvest index. Recently, millets have returned as a viable option to live healthy life and can reduce the incidence of lifestyle diseases viz. Diabetes, Hypertension and Cardiovascular. Barnyard Millets have many nutritional, nutraceuticals and health promoting properties especially the high fiber content. Significance of our research

work in M1 generation phenotypically different characters was noted in Barnyard Millet (Phule Barti-1) with Chemical mutagens like Ethyle Methane Sulphonate (EMS). Observed phenotypic traits are noted viz. dwarf plants, anthocyanin pigment developed plants, plants with maximum height with multiple tillers, sturdy shoot with elongated internodes, fully developed seeds in panicle with curved fingers. Ultimate aim of research work to enhance nutritional parameters with wide consuming acceptability of Barnyard Millets through induced mutation.

Keywords: EMS, Induced Mutation, Phule Barti-1, SA

Abstract-24

Determination of physicochemical parameters and adulteration of multifloral honey by FTIR-ATR spectroscopy

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Abstract: Nowadays there is increase interest towards honey origin and adulteration of pure honey. In this study, physicochemical properties like pH, moisture content, free acidity, diastase activity, reducing sugar and 5-Hydroxymethylfurfural were determined in multifloral honey samples. Natural honey samples were collected during 2020-22 from local honey collecting tribals of Karjat, Ahmednagar district, Maharashtra. Total four adulterants used like glucose, fructose, sucrose and corn syrup. Here, we emphasize the easy, accurate and precise fourier transform infrared and attenuated total reflectance (FTIR-ATR) spectroscopic method for the determination of natural honey and adulterated honey samples. FTIR spectral range of 4000 to 400 cm⁻¹ wavenumbers, however 2934, 1642, 1740, and 800-1200 cm⁻¹ shows predominant wavenumber. Chemometric techniques such as principal component analysis, discriminant analysis and partial least square will be planned for grouping, classification and to differentiation of adulterated honey from natural one. The FTIR-ATR technique with chemometrics provides significant result and better resolution method for detecting adulterated honey.

Keywords: Honey, Adulterants, FTIR-ATR, Physicochemical properties

Abstract-25

Ni-Ferrite an efficient catalyst for synthesis of 3-aryl substituted isoxazole-5-carboxylic acids via one-pot multicomponent reaction

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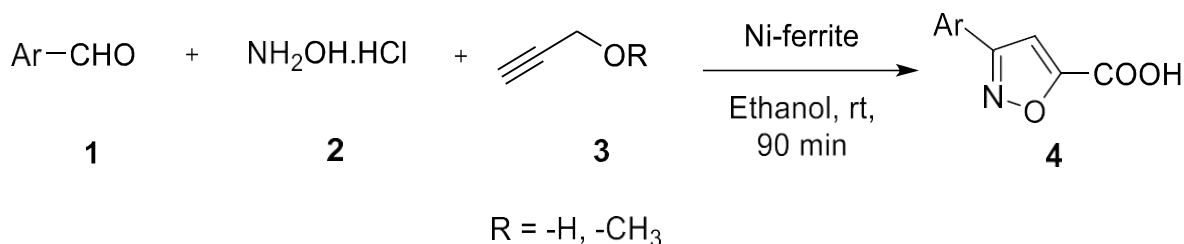
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Abstract: Here, we have reported an efficient one-pot, three-component reaction between various aryl aldehydes, NH₂OH.HCl and propargyl alcohol using Ni-Fe₂O₄ as a nanocatalyst in ethanol at room temperature to afford various 3-aryl substituted isoxazole-5-carboxylic acids in good yields. The advantage of this protocol is that it formed isoxazole derivatives with a free carboxylic acid functionality in one pot which could transform into various other functional groups. This approach has several merits, including a quick reaction time, a wide range of substrates, excellent product yields, and reusable catalysts.



Keywords: Multicomponent; Nickel Ferrites; Isoxazole; Oxidation; Carboxylic acid